

$$a) \quad PV = nRT \Leftrightarrow P = \frac{nRT}{V} = 6 \times 10^5 \text{ Pa}$$

$$V_q = \sqrt{\frac{3k_B T}{m_0}} \Leftrightarrow T = \frac{V_q^2 m_0}{3k_B} = 288.7 \text{ K}$$

B //

$$b-1) \quad \text{isobaric expansion} \Rightarrow \Delta P = 0 \rightarrow P_f = P_i$$

$$PV = nRT$$

$$\frac{T_i}{T_f} = \frac{\frac{P V_i}{nR}}{\frac{P V_f}{nR}} = \frac{V_i}{V_f} = 0.8 \rightarrow A //$$

$$b2) \quad W = - \int P dV \quad \text{Isobaric: } W = -P(V_f - V_i) \quad \left| \rightarrow C // \right.$$

$$\text{Isothermal: } W = - \int \frac{nRT}{V} dV = -nRT \ln\left(\frac{V_f}{V_i}\right)$$

$$b3) \quad T_i = 290 \text{ K}$$

$$\text{Isothermal expansion} \rightarrow \Delta T = 0 \Rightarrow \Delta U = 0$$

$$\Delta U = W + Q \rightarrow \Delta U = 0 \rightarrow W + Q = 0 \rightarrow Q = -W = nT_i \ln\left(\frac{V_f}{V_i}\right) R = 65R \rightarrow A //$$

$$b4) \quad \Delta U = n c_v \Delta T \rightarrow D //$$

$$c) \quad \text{Monatomic} \rightarrow 3 \text{ grados de libertad } c_v = \frac{3}{2} R \rightarrow c_p = c_v + R = 2.5R \rightarrow A //$$